

Tetracaine decreases intracellular Zn²⁺ concentration by inhibiting Zn²⁺ influx in rat thymocytes

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Abstract In this study to examine the cytotoxic property of tetracaine, we cytometrically examined the effect of tetracaine on intracellular Zn²⁺ concentration by the use of FluoZin-3, a fluorescent indicator of intracellular Zn²⁺. Lidocaine was used as a reference drug. The incubation of rat thymocytes with tetracaine decreased the intensity of FluoZin-3 fluorescence while that with lidocaine increased the intensity. The incubation with 10 μM DTPA, a chelator for extracellular Zn²⁺, attenuated the tetracaine-induced decrease in fluorescence intensity. The application of ZnCl₂ augmented FluoZin-3 fluorescence. The augmentation by ZnCl₂ was a temperature-sensitive. Tetracaine attenuated the ZnCl₂-induced augmentation of FluoZin-3 fluorescence. Taken together, the results suggest that tetracaine attenuates membrane Zn²⁺ influx, resulting in a decrease in intracellular Zn²⁺ concentration in rat thymocytes. Although the cells in this study are not targets for actions of local anesthetics, the result may give one clue to explain the difference between the cytotoxicity of local anesthetics since the action of tetracaine on FluoZin-3 fluorescence was opposite to that of lidocaine.

Keywords: Tetracaine • Lidocaine • Zinc • Cytotoxicity

1. Introduction

Local anesthetics such as lidocaine and tetracaine increase intracellular Ca²⁺ concentration by increasing Ca²⁺ release and membrane Ca²⁺ permeability (Johnson et al., 2002; Nishimura et al., 2006). The increase in intracellular Ca²⁺ concentration may induce cell injury and death (Trump and Berezsky, 1995; Orrenius et al., 2003). There is a little information on the effects of local anesthetics on intracellular Zn²⁺ concentration although zinc possesses many physiological roles (Frederickson et al., 2005). Zinc is the second most prevalent trace element and it is involved in the structure and function of over 300 enzymes (Prasad, 1995). Furthermore, it stimulates the activity of approximately 100 enzymes (Sandstead, 1994). Therefore, abnormal increase in intracellular Zn²⁺ concentration may cause some cytotoxic phenomena. Lidocaine at sublethal concentrations increases intracellular Zn²⁺ concentration in absence of

extracellular Zn²⁺ (Nishimura and Oyama, 2010), suggesting intracellular Zn²⁺ release by lidocaine. Since the toxicity of tetracaine is more potent than that of lidocaine (Lambert et al., 1994; Johnson et al., 2002; Yamashita et al., 2003; Werdehausen et al., 2009) and the mixture of lidocaine and tetracaine is used for dermatologic procedures (Alster, 2007 for review), we examined the effect of tetracaine on intracellular Zn²⁺ concentration of rat thymocytes by the use of a flow cytometer with FluoZin-3. Here we describe that the action of tetracaine on intracellular Zn²⁺ concentration may be opposite to that of lidocaine.

Thymocyte is used for a model experiment because of following reasons. (1) Since thymocytes are prepared without any enzymatic treatment for dissociation, cell membranes are remained intact. Intact membranes are important to examine the effect of test compound on membrane transport in single cells. (2) The process of cell death, apoptosis and necrosis, in thymocytes is well defined (McConkey et

al., 1994; Rinner et al., 1996; Winoto, 1997) and Zn²⁺ is one of factors affecting the process of cell death (Raqib et al., 2007; Someya et al., 2009; Haase and Rink, 2009; Wong et al., 2009). (3) Several types of biological substances and chemical compounds (including lidocaine) induce cell death in thymocytes under *in vivo* and *in vitro* conditions (Choyke et al., 1987; Quaglino and Ronchetti, 2001; Nishimura et al., 2006; Gruver and Sempowski, 2008).

2. Materials and methods

2.1. Materials

Tetracaine HCl, lidocaine HCl, NaCl, MgCl₂, KCl, ZnCl₂, NaOH, glucose, dimethyl sulfoxide (DMSO), and 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HPES) were purchased from Wako Pure Chemicals (Osaka, Japan). Diethylenetriamine-N,N,N',N'',N'''-pentaacetic acid (DTPA) was obtained from Dojin Chemical Laboratory (Kumamoto, Japan). Propidium iodide and FluoZin-3 pentaacetoxymethyl ester (FluoZin-3-AM) were products of Molecular Probes Inc. (Eugene, Oregon, USA). Final concentration of DMSO as a solvent in cell suspension was 0.1 % or less. Incubation with DMSO did not affect cell viability during present experiments.

2.2. Animals and cell preparation

This study was approved by the Committee for Animal Experiments in the University of Tokushima (No.05279). Procedure to prepare cell suspension was similar to that previously described (Chikahisa et al., 1996). Thymus glands dissected from ether-anesthetized rats (Wistar strain, Charles River Laboratories, Yokohama, Japan) were sliced at a thickness of about 1 mm with razor under an ice-cold condition. To dissociate single cells, the slices of thymus glands were triturated by gently shaking in chilled normal Tyrode's solution (in mM: NaCl 150, KCl 5, CaCl₂ 2, MgCl₂ 1, glucose 5, HEPES 10, appropriate amount of NaOH to adjust pH at 7.3-7.4) or Ca²⁺-free Tyrode's solution (in mM: NaCl 150, KCl 5, MgCl₂ 3, glucose 5, HEPES 10, appropriate amount of NaOH to adjust pH at 7.3-7.4). Thereafter, Tyrode's solution containing single cells was passed through a mesh (a diameter of 10 μm) to prepare cell suspension (about 5 × 10⁵ cells/ml). The cell suspension was incubated at 36-37°C for 1 h at least before the experiment. Sixteen rats were used in this study. Each result was obtained from

four experiments at least.

2.3. Fluorescence measurements of cellular and membrane parameters

The methods for measurements of cellular and membrane parameters by a flow cytometer (CytoACE-150, JASCO, Tokyo, Japan) and fluorescent probes were similar to those previously described (Chikahisa et al., 1996; Matsui et al., 2008). The fluorescence was analyzed by JASCO software (Ver.3XX, JASCO). Chemicals, except for fluorescent probes, used in this study exerted no fluorescence under our experimental condition.

Propidium iodide at 5 μM was added to cell suspension to assess cell viability. Since propidium stains dead cells, the measurement of propidium fluorescence from cells provides a clue to estimate cell lethality. The fluorescence was measured at 2 min after the application of propidium iodide. Excitation wavelength for propidium was 488 nm and emission was detected at 600 ± 20 nm.

FluoZin-3-AM (Gee et al., 2002) was used as a fluorescent indicator for intracellular Zn²⁺. The cells were incubated with 500 nM FluoZin-3-AM for 60 min before any fluorescence measurements (Hashimoto et al., 2009; Kawanai et al., 2009). FluoZin-3 fluorescence was measured from the cells that were not stained with propidium iodide (Matsui et al., 2008). Excitation wavelength for FluoZin-3 was 488 nm and emission was detected at 530 ± 20 nm.

2.4. Statistics

Values were expressed as mean ± standard deviation of 4 experiments. Statistical analysis was performed by using Tukey's multivariate analysis. A *P* value of < 0.05 was considered significant.

3. Results

3.1. Effects of tetracaine and lidocaine on FluoZin-3 fluorescence of rat thymocytes

As shown in Fig. 1A, tetracaine at 1 mM shifted the histogram of FluoZin-3 fluorescence to a direction of lower intensity, suggesting tetracaine-induced decrease in intracellular Zn²⁺ concentration. Lidocaine at 10 mM increased the intensity of FluoZin-3 fluorescence (Fig. 1A), suggesting lidocaine-induced increase in intracellular Zn²⁺ concentration. The effects of lidocaine and

tetracaine on FluoZin-3 fluorescence were time-dependent. However, their effects attained steady state levels within 60-90 min after the start of application. Therefore, the effects were tested at 90-120 min after the start of drug application. Tetracaine at 0.1 mM or less and lidocaine at 1 mM or less did not affect FluoZin-3 fluorescence. The effect of tetracaine at 0.3-1 mM on FluoZin-3 fluorescence was opposite to that of lidocaine at 3-10 mM (Fig. 1B).

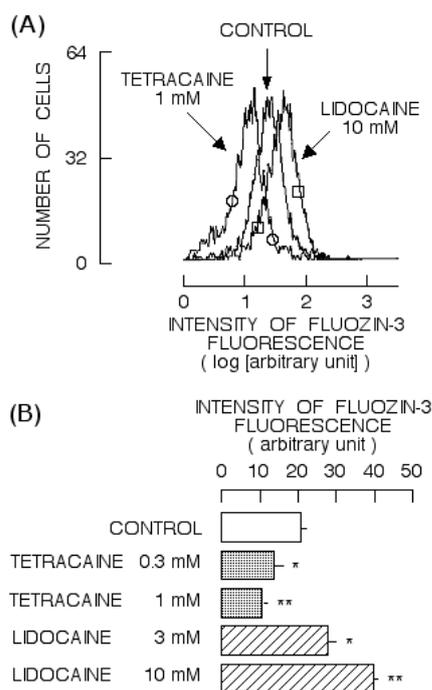


Fig. 1.

Effects of tetracaine and lidocaine on FluoZin-3 fluorescence. (A) Histograms of FluoZin-3 fluorescence monitored from control cells (center histogram) and the cells incubated with 1 mM tetracaine (left with circles) or 10 mM lidocaine (right with squares). Each histogram was constructed with 2000 cells. (B) Mean intensity of FluoZin-3 fluorescence monitored from the cells incubated with 0.3-1 mM tetracaine or 3-10 mM lidocaine. Column and bar indicate mean value and standard deviation of four experiments. Symbols (*, **) show significant changes ($P < 0.05$, $P < 0.01$) to the control.

The concentrations of tetracaine and lidocaine to affect FluoZin-3 fluorescence (Fig. 1B) were sublethal since tetracaine at 3 mM (Fig. 2A) and lidocaine at 30 mM (Nishimura et al., 2006) started to increase cell lethality, respectively. Dose-dependent action of tetracaine on mean intensity of FluoZin-3 fluorescence is summarized in Fig. 2B. Tetracaine at 0.1 mM or more seems to decrease intracellular

Zn^{2+} concentration.

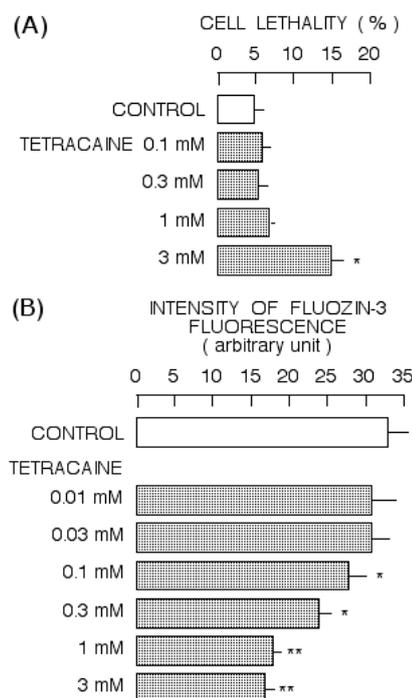


Fig. 2.

Concentration-dependent effects of tetracaine on cell viability and FluoZin-3 fluorescence. (A) Cell lethality in the cells incubated with 0.1-3 mM tetracaine. (B) Concentration-dependent attenuation of mean FluoZin-3 fluorescence intensity by tetracaine. Column and bar indicate mean value and standard deviation of four experiments. Symbols (*, **) show significant changes ($P < 0.05$, $P < 0.01$) to the control.

3.2. Effect of DTPA on tetracaine-induced decrease in intensity of FluoZin-3 fluorescence

The cell suspension contained 200-250 nM zinc derived from cell preparation (Sakanashi et al., 2009). To see if the decrease in Zn^{2+} influx is involved in the tetracaine-induced attenuation of fluoZin-3 fluorescence, the effect of tetracaine was examined in presence of DTPA, a chelator for extracellular Zn^{2+} . The cells were incubated with 10 μ M DTPA for 60 min before applying tetracaine. The incubation of cells with DTPA decreased control level of FluoZin-3 fluorescence and tetracaine at 1 mM further decreased the intensity of FluoZin-3 fluorescence (Fig. 3A). However, the tetracaine-induced decrease in the intensity of FluoZin-3 fluorescence in the presence of DTPA (-27.6 %) was smaller than that under control condition (-43.4 %).

Ca^{2+} competes with Zn^{2+} in some ion transports of membranes (Oyama et al., 1982; Zhuang and Ahearn,

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1996; Sensi et al., 1997). Further experiments were performed under nominal Ca²⁺-free condition where CaCl₂ in Tyrode's solution was replaced with equimolar MgCl₂. As shown in Fig. 3B, the removal of extracellular Ca²⁺ significantly increased the intensity of FluoZin-3 fluorescence. The incubation with 1 mM tetracaine also attenuated the FluoZin-3 fluorescence under nominal Ca²⁺-free condition. The incubation with DTPA significantly reduced control level of FluoZin-3 fluorescence (-63.8 %). In the presence of DTPA, tetracaine did not decrease the intensity of FluoZin-3 fluorescence under nominal Ca²⁺-free condition (Fig. 3B). Results suggest the possibility that tetracaine inhibits Zn²⁺ influx.

3.3. Effect of tetracaine on ZnCl₂-induced augmentation of FluoZin-3 fluorescence

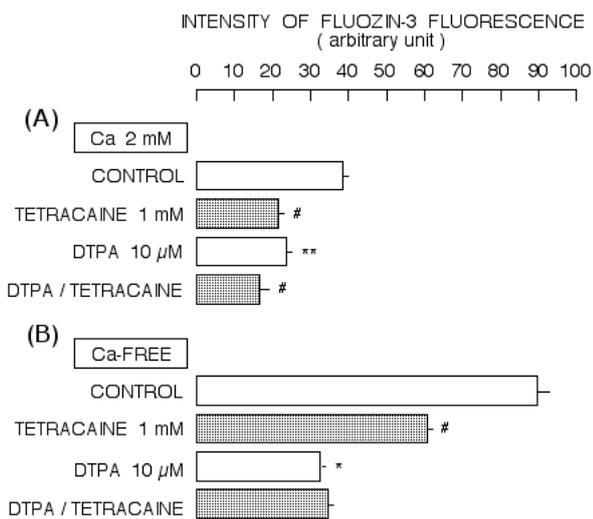


Fig. 3.

Effect of DTPA on tetracaine-induced attenuation of FluoZin-3 fluorescence. (A) The effect under normal Ca²⁺ condition. (B) The effect under nominal Ca²⁺-free condition. Column and bar indicate mean value and standard deviation of four experiments. Symbols (*, **) show significant changes ($P < 0.05$, $P < 0.01$) to respective control. Symbol (#) indicates significant change between pared columns.

To test the possibility, the effect of tetracaine on ZnCl₂-induced augmentation of FluoZin-3 fluorescence was examined. Before the experiments, we examined the effect of ZnCl₂ on the intensity of FluoZin-3 fluorescence under cold condition to see if Zn²⁺ influx is passive. As shown in Fig. 4, the addition of 3 μM ZnCl₂ significantly increased the intensity of FluoZin-3 fluorescence under control

condition (36-37°C) while it was not the case under cold condition (3-4°C). Thus, the augmentation of FluoZin-3 fluorescence by adding ZnCl₂ was not passive.

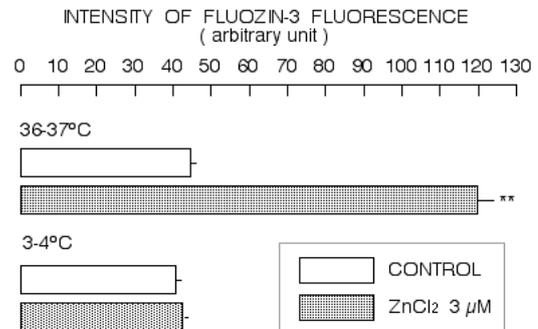


Fig. 4.

Effect of low temperature on the augmentation of FluoZin-3 fluorescence by adding ZnCl₂ into the cell suspension. (Upper panel) The augmentation of FluoZin-3 fluorescence under normal temperature (36-37°C). (Lower panel) Complete attenuation of ZnCl₂-induced augmentation of FluoZin-3 fluorescence under low temperature condition (3-4°C). Column and bar indicate mean value and standard deviation of four experiments. Symbol (**) shows significant change ($P < 0.01$) to the control.

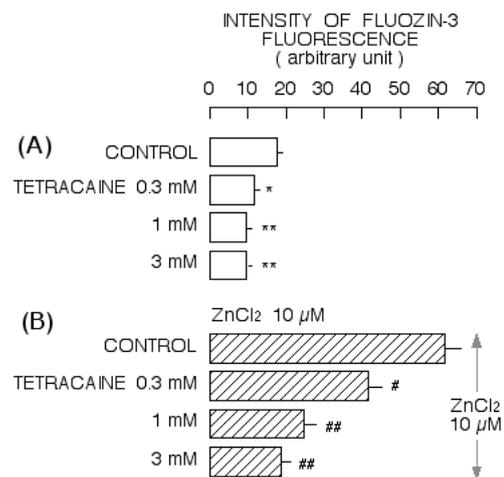


Fig. 5.

Concentration-dependent effect of tetracaine on the augmentation of FluoZin-3 fluorescence by ZnCl₂. (A) The effect of tetracaine on control FluoZin-3 fluorescence. (B) The effect of tetracaine on the augmentation of FluoZin-3 fluorescence by ZnCl₂. Symbols (*, **) show significant changes ($P < 0.05$, $P < 0.01$) to the control without adding ZnCl₂. Symbols (#, ##) indicate significant changes ($P < 0.05$, $P < 0.01$) to the control with adding ZnCl₂.

The cells were incubated with 0.3-3 mM tetracaine for 60 min before adding 10 μ M ZnCl₂ (Fig. 5A). Addition of ZnCl₂ significantly increased the intensity of FluoZin-3 fluorescence under control condition. Tetracaine at 0.3-3 mM attenuated the ZnCl₂-induced augmentation of FluoZin-3 fluorescence in a concentration-dependent manner (Fig. 5B).

4. Discussion

4.1. Tetracaine-induced decrease in intensity of FluoZin-3 fluorescence

The incubation of cells with tetracaine at 0.1 mM or more (up to 1 mM) decreased the intensity of FluoZin-3 (Fig. 2B), suggesting tetracaine-induced decrease in intracellular Zn²⁺ concentration. The incubation with DTPA reduced control level of FluoZin-3 fluorescence and attenuated the tetracaine-induced decrease in FluoZin-3 fluorescence intensity (Fig. 3), indicating the involvement of extracellular Zn²⁺. The cell suspension contained 200-250 nM Zn²⁺ derived from cell preparation (Sakanashi et al., 2009). Zn²⁺ influx seems to maintain intracellular Zn²⁺ level in this preparation. Tetracaine attenuated the ZnCl₂-induced augmentation of FluoZin-3 fluorescence (Fig. 5B). It is reminiscent of the possibility that tetracaine attenuates Zn²⁺ influx, resulting in the decrease in intracellular Zn²⁺ concentration in rat thymocytes. This Zn²⁺ influx seems to be temperature-sensitive because the ZnCl₂-induced augmentation of FluoZin-3 fluorescence was not observed under cold condition (Fig. 4). Such Zn²⁺ influxes were shown in divalent metal transporter 1 (Andrews, 1999; Olivi et al., 2001; Forbes and Gros, 2003; Knöpfel et al., 2005). It may be suggested that tetracaine inhibits Zn²⁺ influx *via* divalent metal transporter 1.

4.2. Toxicological implication

Lidocaine increased the intensity of FluoZin-3

fluorescence (Nishimura and Oyama, 2010) while tetracaine decreased the intensity (Fig. 1). Thus, the action of tetracaine on intracellular Zn²⁺ concentration seems to be opposite to that of lidocaine. Zinc is required in the maintenance and functioning of many proteins/metalloenzymes (Coleman, 1992; Choi and Koh, 1998) and enzymes involved in apoptosis such as endonuclease and caspase are abrogated by physiological concentrations of Zn²⁺ (Gunshin et al., 1997; Giannakis et al., 1991; Perry et al., 1997). If tetracaine decreases intracellular Zn²⁺ concentration, it would increase cell vulnerability to apoptosis (McCabe et al., 1993). It was reported that tetracaine, but not lidocaine, induced apoptosis in rat cortical astrocytes (Lee et al., 2009). Such a difference may be due to the effect on intracellular Zn²⁺ concentration. The mixture of tetracaine and lidocaine is used for dermatologic procedures (Alster, 2007 for review). Such a combination may minimize the change in intracellular Zn²⁺ concentration induced by respective local anesthetics under clinical condition. In preliminary experiment using FluoZin-3 and rat thymocytes, the change in FluoZin-3 fluorescence by the mixture of lidocaine and tetracaine was significantly smaller than those respectively induced by tetracaine and lidocaine (unpublished observation). The clinical efficacy of tetracaine was compared with that of lidocaine by several investigators (Noorily et al., 1995; Stevens et al., 1997; Amiel and Koch, 2007; Bourliias et al., 2009). Furthermore, toxic effects of local anesthetics were also compared in clinical and animal studies (Lambert et al., 1994; Yamashita et al., 2003; McGee and Fraunfelder, 2007; Werdehausen et al., 2009). Although this study may give one insight for such comparisons of local anesthetics, the data would be potentially interesting if the results are supported by the experiments that are relevant for effects of local anesthetics in further study.

Conflict of interest statement

All authors declare that there are no conflicts of interest in this study.

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